

Extraction of *Cecropia Pachystachya* Leaves by Supercritical Carbon Dioxide: Kinetics, Phytochemical Characterization, Antibacterial and Antioxidant Activities

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Abstract

Cecropia pachystachya Trécul is popularly known in Brazil as embaúva or embaúba. Pharmacological activities were reported in extant studies including the development of anxiolytic, antidepressant, hypoglycemic, antioxidant, cardiogenic, sedative, anti-inflammatory, antimicrobial and leishmanicidal properties for aqueous and alcoholic extracts prepared by infusion or maceration of *C. pachystachya* leaves. However, there is a paucity of studies examining extracts of leaves of *C. pachystachya* obtained by supercritical fluid technology. This study evaluated the extraction of *Cecropia pachystachya* oil using supercritical carbon dioxide (CO₂). Extractions were performed at 35 °C, 45 °C, and 55 °C and 14 MPa, 17 MPa, and 20 MPa. The extraction yield, characteristics of the oil, antioxidant and antibacterial potentials were evaluated. The highest extraction yield obtained from *C. pachystachya* leaves was 1.48% using supercritical CO₂ at 55 °C and 20 MPa. The main identified compounds include terpenoids (ursolic and pomolic acids, squalene and β-sitosterol). The most promising results for antibacterial activity relative to *Staphylococcus*

aureus and antioxidant potentials were observed on the extract obtained at 55 °C and 14 MPa. The mathematical Sovová model satisfactorily represented the experimental data.

Keywords: embaúba, Sovová model, supercritical extraction, antimicrobial activity, antioxidant potential

1. Introduction

Cecropia pachystachya Trécul (Urticaceae) is popularly known in Brazil as embaúva, embaúba, or tore and is a tree that is used for ornamental and medicinal purposes. In Brazil is widely used in traditional medicine for treating respiratory infections (Da Silva et al., 2010), hyperlipidemia (Souza & Felfili, 2006), and diabetes (Rodrigues & Carvalho, 2001). It is also used as a diuretic (Rocha et al. 2007). Pharmacological activities were reported in extant studies including the development of anxiolytic, antidepressant (Velázquez et al., 2003; Consolini et al., 2006; Gazal et al., 2014; Ortmann et al., 2016), hypoglycemic (Aragão et al., 2010), antioxidant (Consolini et al., 2006; Farias et al., 2013; Souza et al., 2014; Pacheco et al., 2014), cardiogenic (Consolini & Migliori, 2005; Gazal et al., 2014), sedative (Hikawczuk et al., 1998), anti-inflammatory (Pacheco et al., 2014; Maquiaveli et al., 2014), antimicrobial (Souza et al., 2014), and leishmanicidal (Cruz et al., 2013) properties for aqueous and alcoholic extracts prepared by infusion or maceration of *C. pachystachya* leaves.

Phytochemical surveys already indicated the presence of several groups of flavonoids, tannins (Souza et al., 2014; Ortmann et al., 2016; Ortmann et al., 2017), and catechins (Costa et al., 2011; Cruz et al., 2013). High-performance liquid chromatographic analyses allowed the identification of the flavonoids C-glycosides including isoorientin, isovitexin, orientin, vitexin, and rutin (Costa et al., 2011; Brango-Vanegas et al., 2014), and chlorogenic acid (Brango-Vanegas et al., 2014).

The quality of the vegetal extracts is extremely influenced by the methodology of extraction employed. However, non-optimized processes limit the extractive potential and consequently interfere in the pharmacological activity. Extracts rich in phenolic compounds including flavonoids and tannins are obtained by various solvent extraction techniques and more recently by supercritical fluid technology (Souza-Moreira et al., 2010).

Extraction using supercritical CO₂ fluid technology is attractive due to selectivity, solubility, and mass transfer rates. The selectivity of the extraction process mainly depends on the density of the supercritical CO₂, and this can be altered by varying the process conditions (Subroto et al., 2017). The low latent heat of evaporation in the process and the high volatility of the solvents make it possible to obtain extracts free of toxic residues and to preserve thermally degradable products, thereby obtaining extracts with high quality when compared to the products obtained by conventional techniques (Souza-Moreira et al., 2010; Martinez-Correa et al., 2017). Furthermore, this technique is considered green technology because the solvent is completely removed after the extraction process (Martinez-Correa et al., 2017). However, there is a paucity of studies examining extracts of leaves of *C. pachystachya* obtained by supercritical fluid technology. The objective of the present study is to evaluate the kinetics of supercritical CO₂ extraction, effects of temperature and pressure on

yield and antimicrobial and antioxidant activities, and to perform a preliminary chemical characterization of the compounds present in the extracts.

2. Method

2.1 Plant Material Preparation and Characterization

The plant material was collected at Quatro Pontes, Paraná state, Brazil (24°35'40.56''S; 53°57'46.60''W). The species were identified and a voucher (no. 3000) was deposited at the Herbarium of the West Paraná State University. The material was dried in an oven at 40 °C for 3 days and was ground by a knife grinder. The average size of the particles (720 µm) of *C. pachystachya* leaves was determined by using a Master–Sizer particle analyzer (Malvern Instrument Ltda). The moisture content of the leaves (2.94 ± 0.03 wt%) was obtained by using the gravimetric method to dry sample at 105 °C until the mass stabilized. The density ($1.44 \text{ g}\cdot\text{cm}^{-3}$) of the investigated sample was determined by pycnometer by using helium gas (Micromeritics, AccuPyc model 1330) (Santos et al., 2015).

2.2 Chemicals

Carbon dioxide > 99% (Linde) was used in the supercritical extractions. For the total phenolic concentrations, Folin & Ciocalteu's phenol reagent and gallic acid from Sigma Aldrich were used. Mueller–Hinton agar (Merck) was used in the microbiological analyses. DPPH (2,2-diphenyl-1-picrylhydrazyl), ABTS (2,2'-azino-bis(3-ethylbenzothiazoline-6-sulfonic acid)), TPTZ (2,4,6-tris(2-pyridyl)-s-triazine), Trolox (6-hydroxy-2,5,7,8-tetramethylchromane-2-carboxylic acid), all from Sigma Aldrich, were used for antioxidant analysis.

2.3 *Cecropia pachystachya* Oil Extractions

The extractions were performed in a laboratory-scale unit. The equipment consists of a stainless steel extractor with a capacity of 58 cm³ (with a height of 19.4 cm and a diameter of 1.95 cm), a solvent reservoir, a Syringe pump (Isco, 500D model), and two thermostatic baths. The extractions were performed with pure carbon dioxide (CO₂).

A full factorial experimental 2² design with center points was used to analyze the influence of the independent variables pressure, and temperature, on the extraction yield. The experiments were performed at pressures of 14 MPa, 17 MPa, and 20 MPa and temperatures of 35 °C, 45 °C, and 55 °C. The selection of pressure and temperature conditions allowed an increase in the solvating capacity of the solvent and was based on the equipment limits (Santos et al., 2016).

In each extraction, the vessel was completely loaded with the material. After pressurization, contact between the solvent and material was maintained for 30 minutes, and then the micrometric valve was opened for a solvent mass flow rate of 1.96 g min⁻¹. The oil was collected in an amber vessel, and its mass (% w/w) was determined at time intervals of 10 min (up to 150 min of extraction). The yields were calculated as the ratio of the extracted oil mass to the initial *C. pachystachya* leaves mass. The experimental data were statistically analyzed using Statistica™, version 7.0 software (Statsoft) at the 95% confidence level.

2.4 Kinetics of the *C. pachystachya* Oil Extraction

The kinetic curves were described by a second-order model proposed by Sovová (StatSoft, 2007). The model considers two mass transfer mechanisms and three different extraction stages. In the first stage, the extraction of directly exposed oil (bioactive compounds) in the fluid phase occurs. This step depends on the solubility in the fluid phase and is characterized by a linear curve with a slope close to the solubility value of the oil in the solvent (Silva et al., 2015). Subsequently, a decrease in the extraction rate occurs at which the fraction of oil on the surface of the cells shows indications of being exhausted. Additionally, the extraction of a low amount of accessible oil occurs, and this is controlled by an internal diffusion mechanism (diffusion mass transfer mechanisms in conjunction with the convection). In the third stage, the extraction curve becomes almost linear at an extraction rate that is significantly lower than that in the first period (Nelder & Mead, 1965; Liu et al., 2013; Silva et al., 2015; Lopes et al., 2020). The solubility was calculated from the slope of the linear part of the general extraction curves. The parameters r , Z , and W , were adjusted as previously reported (Santos et al., 2016).

2.5 Phytochemical Characterization

2.5.1 GC-MS Analysis

Chemical identification was performed on GC-MS QP2010 SE (Shimadzu). Specifically, 10 μL of the samples were diluted with 990 μL of dichloromethane (Anidrol) before injecting the same into a SH-RTx-5MS column (Shimadzu, 5% phenyl-methylsiloxane, 30 m \times 0.25 mm id, 0.25 μm) by using an auto-sampler (Shimadzu AOC-20i). Helium was used as the carrier gas at a flow rate of 1.0 ml min^{-1} with a split ratio of 2:1, and the volume of the injected sample was 1 μL . The column temperature was initially programmed at 40 $^{\circ}\text{C}$ and heated at 6 $^{\circ}\text{C min}^{-1}$ to reach the final temperature of 300 $^{\circ}\text{C}$. It remained at this temperature for 10 min. The total time of GC-MS analysis was 53.3 min. The injector and the GC-MS interface temperatures were maintained at 250 $^{\circ}\text{C}$. Mass spectra were recorded at 70 eV with a mass range from m/z 35 to 550 amu.

2.5.2 Total Phenolic Content (TPC)

The TPC of the oils was determined by using the Folin-Ciocalteu methods described previously (Singlenton & Rossi, 1965). The standard calibration (between 500 and 15.625 $\mu\text{g mL}^{-1}$) curve was plotted by using gallic acid given the following equation for the line $y = 0.0064x + 0.1135$ with a correlation coefficient of $R^2 = 0.9932$. All determinations were performed thrice, and the total phenolic content was expressed as $\mu\text{g gallic acid equivalent/g}$ of oil ($\mu\text{g}_{\text{GAE}} \text{g}_{\text{oil}}^{-1}$).

2.6 Antioxidant Activity

2.6.1 DPPH Radical Assay

The DPPH method is based on the quantification of the oil, or extract, concentrations necessary to reduce the initial concentration of the radical DPPH by 50% by the hydrogen and electron donation (Miguel, 2010). The radical scavenging capacities of each oil in

different concentrations (100–1000 $\mu\text{g mL}^{-1}$ in ethanol) were estimated according to the method of Choi et al. (2002). Negative control was prepared by mixing 1 mL of DPPH 0.3 mM with 2.5 mL of ethanol. The mixtures were shaken vigorously for 1 min and left to stand at room temperature (37 °C) in the dark for 30 min. Thereafter, the absorbance for the sample was measured at 517 nm. Essential oils and plant extracts are considered to have antioxidant activity when their IC_{50} value is less than 250 $\mu\text{g mL}^{-1}$ (Czaikoski et al., 2015).

2.6.2 FRAP—Ferric Reducing Antioxidant Analysis

The FRAP analysis was performed using the methodology of Santos et al. (2016) at 595 nm using the FRAP reagent as a blank. A standard calibration (100–2000 μM) curve was plotted by using ferrous sulfate. All determinations were performed in triplicate, and the antioxidant capacity was expressed as $\mu\text{mol of Fe}^{2+}$ per g of oil.

2.6.3 ABTS⁺⁺ Free Radical Scavenging Assay

The ABTS activity was carried out based on a method developed by Santos et al. (2016). The antioxidant activity of the samples was calculated through the range of the dose-response curve of Trolox reagent (100–2000 μM) and presented as $\mu\text{mol of Trolox}$ per g of oil. All reactions were carried out in triplicate.

2.7 Antibacterial Activity

Standardized bacterial strains of *Staphylococcus aureus* (ATCC 25922), *Escherichia coli* (ATCC 25923), and *Pseudomonas aeruginosa* (ATCC 27853) were transferred to sterile 0.9% sodium chloride until a transmittance of 25% was obtained at a wavelength of 580 nm by using a colorimeter. An inoculum with 1.0% in agar was prepared from this suspension by adding 1 mL of suspension into 100 mL of sterile Muller-Hinton agar at 46–48 °C (USP, 2011).

The antibacterial activity was assessed by an agar well diffusion method (CLSI, 2009). This consists of the preparation of plates containing one layer (20 mL) of sterile solid medium and another layer (5 mL) of culture medium that is previously inoculated with sensitive microorganisms. After solidification, five holes were created on the culture medium with a sterile perforator in which two of the holes are impregnated with the *C. pachystachya* oils extracted at a concentration of 30 mg mL^{-1} , and two of the holes are impregnated with antibiotics Amoxicillin and Neomycin Sulfate at a concentration of 10 mg mL^{-1} . The fifth hole was used a negative control (sterile 0.9% sodium chloride). The plates were incubated in a bacteriological oven at 35 ± 2 °C for 24 h (USP, 2011), and the plates were read with a caliper (ANVISA, 2010). All tests were performed thrice.

2.8 Statistical Analysis

The significant differences between means of the experimental results were submitted to variance analysis (ANOVA) and compared by performing Tukey's test ($p < 0.05$) by using Statistica 7.0 software (Statsoft, USA).

3. Results and Discussion

3.1 Extraction Yield

The experimental conditions used for the oil extraction of *C. pachystachya* leaves, CO₂ density, oil solubility (*S*), and yield obtained for each condition are shown in Table 1. The *Y_s* values were calculated from the linear part of the extraction curves of the *C. pachystachya* leaves.

Table 1. Experimental conditions and results for the oil extraction yields of *C. pachystachya* leaves by using supercritical CO₂

Tests	T (°C)	P (MPa)	Solvent density ^a (g cm ⁻³)	<i>S</i> ^b 10 ⁻³ (g cm ⁻³)	Time (min)	Yield (wt%)
A	35	14	0.801	0.60	150	0.95
B	35	20	0.865	1.13	150	1.35
C	55	14	0.618	0.60	150	0.65
D	55	20	0.754	1.40	150	1.48
E	45	17	0.775	0.94	150	1.26 ± 0.004 ^c

Note. ^aCO₂ density calculated according to NIST (2015). ^bOil solubility in the solvent. ^cAverage value ± standard deviation (*n* = 3)

Concerning the temperature and pressure conditions investigated in the experimental design, the pressure was the only variable that exhibited a significant effect (*P* < 0.05). It obtained a maximum increase of approximately 127% (0.65–1.48 wt%) for the yield with an increase in the extraction pressure from 14 MPa to 20 MPa at 55 °C. This was observed in previous studies (Ribas et al., 2014; Santos et al., 2016; Santos et al., 2017). Thus, an increase in pressure leads to increases in the solvating power of CO₂ and results in a higher extraction yield (Palvic et al., 2017). The temperature did not exhibit a statistically significant influence on the extraction yield at the highest pressure that was used. However, the results suggest a positive effect on this variable. At low pressure (14 MPa), an increase in temperature decreases the yield (0.95 at 35 °C and 0.65 at 55 °C, almost 30% of decrease). Whereas at higher pressure (20 MPa), an increase in temperature slightly increases the yields (1.35 at 35 °C and 1.48 at 55 °C). Given the opposing effects of vapor pressure and solvent density and on solubility, the effect of temperature is a complicated parameter. This phenomenon is due to the retrograde solubility usually encountered with supercritical CO₂, which is due to opposite effects of CO₂ density and solute vapor pressure, where the temperature increases with decreases in the solvent density and on solubility (Hatami et al., 2014). A similar effect was reported for the extraction of *Lupinus albus* Hook. & Arn. roots (Conforti et al., 2017), *Humulus lupulus* L.

hops (Kupski et al., 2017), and *Salvia officinalis* L. (Palvic et al., 2017).

The chemical compounds were identified in the oil of *C. pachystachya* extracted with CO₂ (Appendix). The samples presented qualitative similarities in their compositions although the changes in temperature and pressure altered the percentage of the components (Table 2).

Table 2. Chemical composition of the oil extracted from *C. pachystachya* leaves by using supercritical CO₂

Relative area (%)	Tests				
	A	B	C	D	E
Ursolic acid	30.35	46.83	30.68	39.43	39.46
Squalene	15.76	13.29	18.49	16.00	12.92
Pomolic acid	12.76	6.88	15.39	14.53	7.79
β-sitosterol	3.71	6.25	6.12	0.31	8.13
Vitamin E	5.84	3.68	6.90	4.21	4.61
α-amyrin	2.16	1.69	1.81	2.14	2.26
Catechin	4.49	0.91	1.49	5.29	2.40
Isovitexin	0.29	0.14	0.34	0.19	0.14
Isoorientin	0.25	0.34	0.35	0.31	0.43
Chlorogenic acid	6.58	10.9	4.00	5.97	9.55
Others	17.81	9.09	14.43	11.62	12.31

Note. A: 35 °C x 14 MPa, B: 35 °C x 20 MPa, C: 55 °C x 14 MPa, D: 55 °C x 20 MPa, E: 45 °C x 17 MPa

It can be mentioned that approximately 20,000 triterpenes have already been isolated and identified from several medicinal plants, with the tetracyclic and pentacyclic compounds being the most extensively investigated about their pharmacological activities (Yang et al., 2020). Among the triterpenes, squalene it is the simplest representative, being acyclic (Shanmugam et al., 2012). The terpenes β-sitosterol (Rivera-Mondragón et al., 2017), ursolic and pomolic acids, and α-amyrin (Hikawczuk et al., 1998; Rivera-Mondragón et al., 2017), previously found in extracts of *C. pachystachya* leaves, have been identified in a significant

concentration in our oils. The high concentration of these compounds can be justified by their non-polar characteristics, similar to supercritical CO₂ that is a non-polar solvent.

The aqueous and alcoholic extracts obtained from *C. pachystachya* are widely known for their high composition of flavonoids that include catechin (Costa et al., 2011; Cruz et al., 2013), isoorientin and isovitexin (Costa et al., 2011; Cruz et al., 2013; Brago-Vanegas et al., 2014). Additionally, quantification of the major phenolic compounds in the leaves of *C. pachystachya* showed that chlorogenic acid is the major compound in the aqueous extract of this species (Brago-Vanegas et al., 2014). Despite high content of flavonoids in *C. pachystachya*, the content in oils indicates that only a small part of flavonoids was extracted and much larger part remained in residuum. So the polarity of this solvent does not favor the extraction of flavonoids.

3.2 Kinetics and Mathematic Modeling of the Extraction

The apparent (0.775 g cm⁻³) and real (1.447 g cm⁻³) density values obtained allowed the determination of the porosity of the bed (0.812). The characteristics necessary for the implementation of the mathematical Sovová model were as follows: Initial concentration of oil in the inert solid = 0.015 g g⁻¹; Bed density = 0.238 g cm⁻³, and solid mass in an oil free base = 11.528 g. The experimental conditions, solubility, and adjustable parameters are shown in Table 3.

Table 3. Adjustable parameters for the Sovová model

Tests	Z	W 10 ⁻¹	r	S 10 ⁻³ (g _{oil} g _{solvent} ⁻¹)	t _{CER} (min)	t _{FER} (min)	k _{Fa} (min ⁻¹)	k _{sa} 10 ⁻³ (min ⁻¹)	R ²
A	12.65	0.64	0.69	0.60	4.69	94.93	1.76	1.67	0.991
B	12.84	0.70	0.69	1.13	2.76	50.95	0.41	1.53	0.993
C	2.04	0.50	0.69	0.60	34.90	141.08	0.08	1.01	0.986
D	4.30	2.11	0.69	1.40	5.72	51.93	0.18	5.33	0.994
E	12.61	1.30	0.69	0.94	2.88	68.41	0.52	3.28	0.998

Note. A: 35 °C x 14 MPa, B: 35 °C x 20 MPa, C: 55 °C x 14 MPa, D: 55 °C x 20 MPa, E: 45 °C x 17 MPa. Z, dimensionless parameter of Sovová model; W, dimensionless parameter of Sovová model; r, easily accessible oil mass; S, solubility; t_{CER}, time at which the extraction of the oil from the inside of particles starts; t_{FER}, time at which the extraction of easily accessible solute ends; k_{Fa}, solvent-phase mass transfer coefficient; k_{sa}, solid-phase mass transfer coefficient; R², coefficient of determination

The parameter r (0.69) indicates the existence of a high oil level that is easily accessed by the solvent and provided by the milling process. Its value is constant since the raw material was subjected to the same treatment before the extractions. The Z and W coefficients are proportional to the mass transfer parameters in the solvent phase (k_{Fa}) and the solid phase (k_{sa}), respectively. Values were observed in the same order of magnitude for k_{Fa} and k_{sa} in the extraction of Candeia oil by using pressurized CO_2 (Ribas et al., 2014). As observed, the k_{Fa} values exceed the k_{sa} values, and thus demonstrate the higher yield during easy-to-reach oil extraction (Silva et al., 2015).

According to the model, the extraction kinetics are divided into three periods with mass transfer. The first period (t_{CER}) is fast (2.76–34.90 min) because it refers to the oil with the most easily accessed to the extraction. In the intermediate period (t_{FER}), the extraction rate decreases due to the exhaustion of the oil that is readily available until the beginning of the third period in which the extraction rate is limited by the internal diffusion mechanism (Santos et al., 2017) with low mass transfer coefficients in the solid phase k_{sa} ($1.01 \cdot 10^{-3}$ to $5.33 \cdot 10^{-3} \text{ min}^{-1}$) due to the difficulty in extracting the oil contained within the intact cells.

The mathematical modeling indicated a good fit ($R^2 > 0.986$) in all experimental conditions investigated. The values reveal that Sovová's model properly represented the *C. pachystachya* oil extraction kinetic behavior in all the cases. The experimental results of extraction kinetics and Sovová's model are shown in Figure 1.

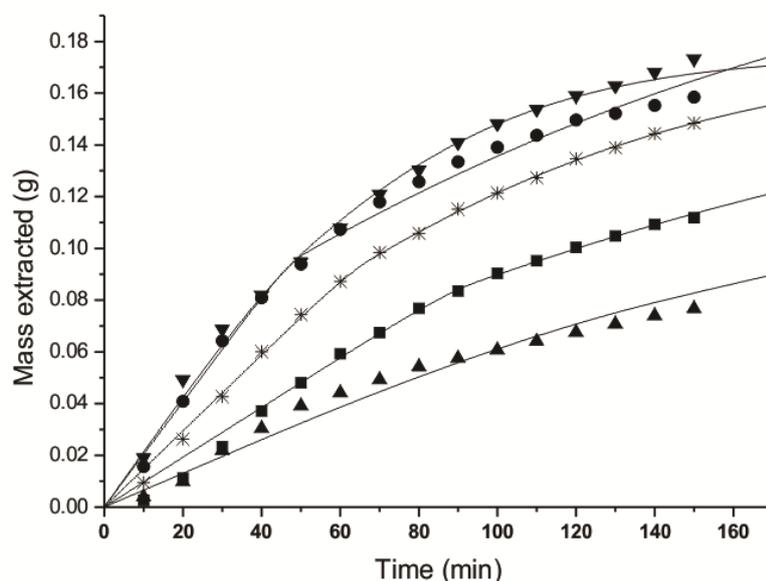


Figure 1. Experimental and modeled kinetic curves for the extraction of *C. pachystachya* oil by supercritical CO_2 extraction using the Sovová model

Note. ▼: 55 °C x 20 MPa, ●: 35 °C x 20 MPa, *: 45 °C x 17 MPa, ■: 35 °C x 14 MPa, ▲: 55 °C x 14 MPa

3.3 Total Phenolic Content, Antibacterial and Antioxidant Activity

The low TPC in extracts ($61.07 - 72.37 \mu\text{g}_{\text{GAE}} \text{g}_{\text{oil}}^{-1}$) proves that only a small part of flavonoids was extracted (Table 4). In general, extracts obtained with supercritical CO_2 present low extraction of TPC due to the non-polar capacity of this solvent that does not favor the extraction of phenolic compounds, a substance of a generally polar character, and responsible for the antioxidant activity (Dias et al., 2017).

Table 4. TPC, antibacterial activity against *S. aureus*, and antioxidant activity of *C. pachystachya* oils extracted with supercritical CO_2

Extracti on	TPC (μg_{GAE} $\text{g}_{\text{oil}}^{-1}$)	Antibacterial activity ^a	Antioxidant activity		
			IC ₅₀ (μg mL^{-1})	FRAP ($\mu\text{mol}_{\text{Fe}^{2+}}$ $\text{g}_{\text{oil}}^{-1}$)	ABTS ($\mu\text{mol}_{\text{Trolox}}$ $\text{g}_{\text{oil}}^{-1}$)
A	64.50 ± 1.42	7	249.85 ± 1.28	266.85 ± 0.01	4177.77 ± 0.07
B	63.10 ± 1.32	6	246.45 ± 1.61	266.57 ± 0.01	3900.00 ± 0.12
C	72.37 ± 1.42	9	196.39 ± 1.31	381.14 ± 0.02	5288.88 ± 0.70
D	64.09 ± 0.73	6	234.49 ± 0.62	326.14 ± 0.16	3900.00 ± 0.15
E	61.07 ± 0.65	7	359.23 ± 0.73	191.14 ± 0.00	3122.22 ± 0.32
NC	-	0	0	0	0
Amo	-	15	-	-	-
NS	-	15	-	-	-

Note. TPC and antioxidant activity values expressed in mean \pm standard deviation ($n = 3$).

^aAntibacterial activity values expressed in terms of the mean of the inhibition zone (mm) ($n = 3$). One-way ANOVA post hoc Tukey test A: $35^\circ\text{C} \times 14 \text{ MPa}$, B: $35^\circ\text{C} \times 20 \text{ MPa}$, C: $55^\circ\text{C} \times 14 \text{ MPa}$, D: $55^\circ\text{C} \times 20 \text{ MPa}$, E: $45^\circ\text{C} \times 17 \text{ MPa}$, NC: Negative control, Amo: Amoxicillin, NS: Neomycin Sulfate

The oils extracted from *C. pachystachya* leaves were evaluated for their antioxidant capacity. The DPPH method revealed that the temperature and pressure employed in the extraction

influenced the antioxidant capacity, which ranged from 196.39 $\mu\text{g mL}^{-1}$ (55 °C x 14 MPa) to 359.23 $\mu\text{g mL}^{-1}$ (45 °C x 17 MPa). These results are significant when compared to other studies evaluating the DPPH method from *Eremanthus erythropappus* supercritical CO₂ extract that found IC₅₀ of 716 $\mu\text{g mL}^{-1}$ (Santos et al., 2017), and spent coffee extract who reported IC₅₀ values of 478 to 2369 $\mu\text{g mL}^{-1}$ (Andrade et al., 2012).

In the tests using the FRAP method, the antioxidant potentials of extracts were estimated for their ability to reduce the TPTZ-Fe (III) complex to a TPTZ-Fe (II) complex. The FRAP reducing values were in the range of 191.14-381.14 $\mu\text{mol}_{\text{Fe}^{2+}} \text{g}_{\text{oil}}^{-1}$. Again, the most effective condition for the recovery of reducing substances in the FRAP assay was observed at 55 °C and 14 MPa. In contrast, reducing capacity of 470 $\mu\text{mol}_{\text{Fe}^{2+}} \text{g}_{\text{oil}}^{-1}$ by FRAP method was observed for the supercritical CO₂ extract from *Populus nigra* (Kús et al., 2018). So, it can be showed lower antioxidant activity by FRAP method for the extracted oil.

According to ABTS method, based on the capacity for reducing the chromophore radical ABTS^{•+}, the antioxidant potential varies from 3122.22 (45 °C x 17 MPa) to 5288.88 (55 °C x 14 MPa) $\mu\text{mol}_{\text{Trolox}} \text{g}_{\text{oil}}^{-1}$. These results showed greater antioxidant potential by ABTS method compared to the results reported to candeia wood oil (232.11 to 405.48 $\mu\text{mol}_{\text{Trolox}} \text{g}_{\text{oil}}^{-1}$) obtained by supercritical CO₂ (Santos et al., 2016). The best results for TPC and antioxidant potential were observed for "C" extract obtained at high temperature (55 °C) and low pressure (14 MPa), thus although the temperature does not show statistically significant on the extraction yield, the biological results obtained for "C" show the significant effect of temperature.

This study involved an evaluation of the *in vitro* antibacterial activity against *P. aeruginosa*, *E. coli*, and *S. aureus*, but only the last species exhibited inhibition to the concentration used (30 mg mL⁻¹). There is no consensus on the acceptable level of inhibition for natural products when compared to antibiotic standards since the material is a mixture of compounds and not a single compound in isolation (Almeida et al., 2014; Waltrich et al., 2015).

A study evaluating the antimicrobial activity of the aqueous and alcoholic extracts of *C. pachystachya* leaves against *S. aureus* showed an inhibition zone of 7 mm for the alcoholic extract (at 100 mg mL⁻¹) (Costa & Hoscheid, 2018). This inhibitory action was attributed to flavonoids and TPC, since they are well established as antimicrobial agents (Babii et al., 2016; Tsou et al., 2016; Echeverría et al., 2017). However, when compared with this work, we observed the formation of a greater inhibition zone (9 mm) in a lower concentration (30 mg mL⁻¹) by the same method (agar diffusion method). This fact makes us understand that the main antimicrobial activity found in "C" extract (55 °C x 14 MPa) is not due to the unique presence of flavonoids, which showed low content by GC-MS and TPC.

Extant studies have shown discrete effects of β -sitosterol on *S. aureus* (Chandra & Saklani, 2017; Nyawai et al., 2017), and the synergic action of ursolic acid combined with penicillin (Wenyi et al., 2010) and β -lactams (Catteau et al., 2017) on drug-resistant *S. aureus*. Hence, the major compound ursolic acid may be acting synergistically with flavonoids and β -sitosterol to employ the antimicrobial activity observed.

4. Conclusion

The results of the study indicated that the oil extraction of *C. pachystachya* with supercritical CO₂ provided satisfactory extract yields in 150 min. The best results were obtained by applying a higher pressure and temperature (20 MPa and 55 °C) investigated in the study with a yield of 1.48 wt%. The examined experimental conditions revealed that only one pressure exhibited a statistically significant effect without extraction yield, however, the biological results obtained show the significant effect of temperature. The oil obtained by applying different conditions did not reveal significant differences concerning its qualitative composition. The major identified compounds were ursolic and pomolic acids. The best results for TPC, antibacterial activity relative to *Staphylococcus aureus*, and antioxidant potentials were observed on the extract obtained at high temperature (55 °C). For extraction kinetics, the Sovová model provided an adjustment with experimental data.

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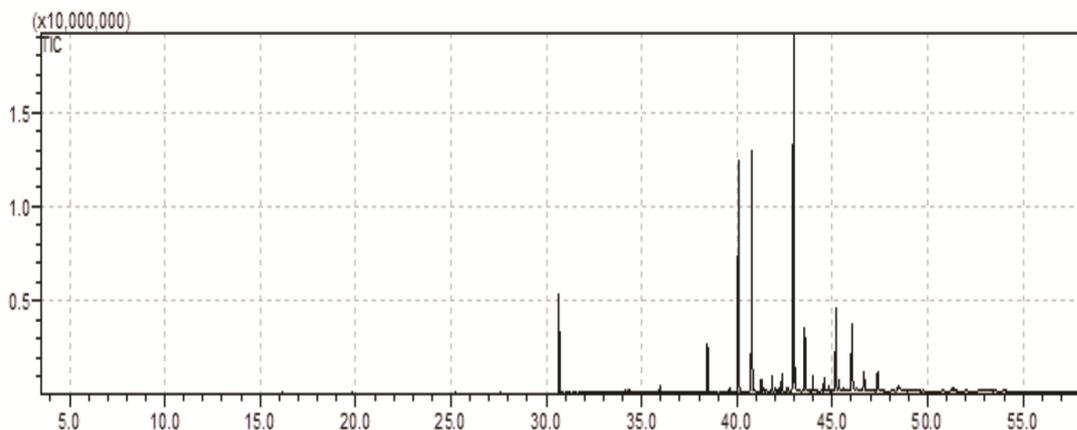
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Appendix



Appendix. Screening by GC-MS of *Cecropia pachystachya* oil

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